## Total Pages - 2

## II-Sem/COMMON/2019(S)(New)

## ENGINEERING CHEMISTRY

(Theory: 2(b))

Full Marks: 80

Time: 3 hours

Answer any five questions including Q. Nos. 1 & 2

Figures in the right-hand margin indicate marks

1. Answer all questions:

 $2 \times 10$ 

- (a) What are ore and mineral?
- (b) Define galvanisation and electrorefining?
- (c) Calculate the normality of H<sub>2</sub>SO<sub>4</sub> solution containing 14.7 grams of solute in 3 litres of solution?
- (d) What are insecticides? Give two examples?
- Define isotopes and isotones.
- What are primary fuels? Give examples two geseous primary fuels?
- (g) What are saturated hydrocarbons? Give examples.
- (b) What are acid radicals? Give examples of two divalent monoatomic acid radicals?
- (i) How Na<sub>2</sub>CO<sub>3</sub> removes permanent hardness?
- (j) Calculate equivalent weight of MgCO<sub>3</sub> and H<sub>3</sub>PO<sub>4</sub>?
- 2. Answer any six questions:

 $5 \times 6$ 

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- (v) Write the failure of Rutherford's Atomic Model?
- (b) Explain the formation of H<sub>2</sub>O molecule?
- What is concentration of ore? Write the magnetic separation process?
- (d) Give comparison between various liquid fuels? http://www.sctevtonline.com
- (e) Write the composition and use of polyvinyl chloride?
- (f) Write the IUPAC names and structures of following compounds:
  - (i) CH<sub>3</sub>C(Br<sub>2</sub>)CHCHCH(C<sub>2</sub>H<sub>3</sub>)CH<sub>3</sub>
  - (ii) CH<sub>2</sub>CHCH<sub>2</sub>CCH
  - (iii) Isoprene
  - (iv) BHC
  - (v) 2-bromo-5-chloro-3, 4-diethyl hexane.
- (g) 40 gms of caustic soda is dissolved in water to prepare 5 litres of its solution, having density 1.02 gm/cc. Calculate the normality, molarity and molality of the solution?

(Turn Over)

(2)

(h) What are bio-fertilizers? Write uses of various such fertilizers?

|    | (i) If 6.9 gms of K <sub>2</sub> CO <sub>3</sub> is present in 0.5 litres of its solution, then calculate normality and pH of the solution?                             | the           |
|----|---|---------------|
| 3. | Explain Bohr's Atomic Model? How this model rectified the defect of Rutherfor Atomic Model?   | rd's<br>6 + 4 |
| 4. | Make a comparison between Arrhenius Theory and Lowry-Bronsted theory?   | ? 4           |
|    | (b) Explain the atmospheric corrosion?  | 5             |
| 5. | (4) What is electrolysis? Explain the process of electroplating?  | 1 + 5         |
|    | (b) Write the limitations of Lewis theory for acid and base?  | 4             |
| 6. | (a) Write the composition and uses of brass, alnico and duralumin?  | 6             |
|    | (b) Write the uses of naphthalene and benzoic acid?   | 2 + 2         |
| 7. | (a) 21.2 gms of sodium carbonate is dissolved in water to prepare 2 litres of<br>solution, having density 1.01 gm/cc. Calculate the normality, molality of<br>solution? | its<br>the    |
|    | (b) Explain ion-exchange process of water softening?  | (             |

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